

*Note Added in Proof:* Recent detailed analysis of the multigroup diffusion calculations indicates that the good agreement between calculation and experiment is partly fortuitous. The effect of the diffusion approximation as applied to the  $D_2O$  is to underestimate both the ratio of  $U^{238}$  fissions to thermal fissions and the resonance escape probability. Further study of these effects is continuing.

## REFERENCES

1. See, e.g., A. M. WEINBERG AND E. P. WIGNER, "The Physical Theory of Neutron Chain Reactors, Univ. of Chicago Press, Chicago, 1958; or S. GLASSTONE AND M. C. EDLUND, "The Elements of Nuclear Reactor Theory," Van Nostrand, New York, 1952.
2. See, e.g., Y. GIRARD *et al.*, Geneva Conference Paper P/336, 1958; and D. W. HONE, *et al.*, Geneva Conference Paper P/212, 1958.
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## Corrosion of Irradiated Uranium Alloys

The effect of nuclear radiation on some corrosion-resistant uranium alloys has been described in previous papers (1, 2). Essentially, we learned that small radiation dosages effectively destroyed the corrosion resistance of metastable uranium alloys heat-treated for corrosion resistance.

Accordingly, the fuel for the EBWR, U-5% Zr-1½% Nb, was heat-treated for dimensional stability under irradiation. This treatment comprised heating at 825°C, quenching at 640°C, holding at 640°C for 23 hr, and air cooling. In this condition the fuel material has a corrosion rate<sup>1</sup> at 260°C in initially pure water of 9470 mg/cm<sup>2</sup>/day. For comparison, the corrosion rate for unalloyed uranium under the same conditions is about 64,000 mg/cm<sup>2</sup>/day.

At a scheduled shutdown of the EBWR a subassembly was removed and sections taken from one of the plates for corrosion testing. Burnup varied with position in the plate and was determined by radiochemical analysis. The sections were exposed to initially pure water at 260°C-270°C for approximately one day. Results are summarized in Table I.

The results are interesting and surprising. Because of the nature of the work, the results reported here (as well as most of those in previous studies) are necessarily based on one or two samples at each level of burn-up. However, the trend observed appears to be definite enough to warrant publication. At the present time we have no explanation for

<sup>1</sup> The rates discussed in this letter are based on original exposed area. True rates would therefore be higher but the quoted rates are valid for purposes of comparison.

TABLE I  
CORROSION OF IRRADIATED U-5% Zr-1½% Nb AT  
260°-270°C AS A FUNCTION OF BURNUP

Sample No.	Burnup, Total a/o	Corrosion Rate, mg/cm <sup>2</sup> /day
	0.000	9470
1	0.005	2500
15	0.009	2780
13	0.017	2310
12	0.024	1880
10	0.088	1890
6	0.140	5160

the improvement in corrosion resistance or the apparent minimum in corrosion rate as a function of burn-up.

This study is continuing and results will be published as they become available.

The samples and burnup data were furnished by C. F. Reinke of the Metallurgy Division.

## REFERENCES

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## Measurements of the Transport Mean Free Path of Thermal Neutrons in Beryllium as a Function of Temperature

In order to calculate temperature effects on reactors it is important to know the variation of the transport mean free path of thermal neutrons with temperature. K. S. Singwi and L. S. Kothari (1) have calculated this variation for different crystalline materials. In an attempt to confirm these calculations experimentally, we investigated the variation with temperature of the transport mean free path,  $\lambda_{tr}$ , of thermal neutrons in beryllium using the pulsed-neutron technique (2). The decay constant,  $\lambda$ , of the fundamental mode of the thermal neutron density in a beryllium assembly 20 by 20½ by 20½ in. (buckling,  $B^2 = 1.05 \times 10^{-2} \text{ cm}^{-2}$ ) was measured for 10 different temperatures of the beryllium, ranging from -46° to +511°C.

The reciprocal thermal-neutron lifetime,  $\lambda_a = 288 \pm 60 \text{ sec}^{-1}$ , and the room-temperature diffusion cooling constant,  $C = 1.1 \pm 1 \text{ cm}^2$ , for this beryllium had been previously measured (3). The relation between the decay constant  $\lambda$

<sup>2</sup> Operated for United States Atomic Energy Commission by the University of Chicago.